

(1) Publication number:

0 129 947

(12)

#### **EUROPEAN PATENT APPLICATION**

(21) Application number: 84300611.5

(22) Date of filing: 31.01.84

(5) Int. Cl.4: **C** 07 **J** 1/00 C 07 **J** 41/00, A 61 K 31/565

- (30) Priority: 19.05.83 GB 8313921
- (43) Date of publication of application: 02.01.85 Bulletin 85/1
- (84) Designated Contracting States: AT BE CH DE FR GB IT LI LU NL SE
- (7) Applicant: WORLD HEALTH ORGANISATION

CH-1211 Geneva 27(CH)

Inventor: Archer, Sidney Department of Chemistry Renssalaer Polytecnic Institute Troy New York 12181(US)

- (72) Inventor: Diczfalusy, Egon Swedish Medical Research Council Karolinska Sjukhuset S-104 01 Stockholm 60(SE)
- (72) Inventor: Fried, Josef Department of Chemistry The University of Chicago 5735 South Ellis Avenue Chicago Illinois 60637(US)
- (72) Inventor: Benagiano, Guiseppe **World Health Organisation** CH-1211 Geneva 27(CH)
- (72) Inventor: Crabbe, Pierre 7 East Burnam Road Columbia Missouri 65201(US)
- (72) Inventor: Djerassi, Carl **Department of Chemistry Stanford University** Stanford California 94305(US)
- (74) Representative: Arthur, Bryan Edward et al, Withers & Rogers 4 Dyer's Buildings Holborn London EC1N 2JT(GB)

(54) Contraceptive compositions based on esters of levo-norgestrel.

5) A small group of esters of levo-norgestrel have been found to be unexpectedly effective as long term contraceptive agents. The esters are:

levo-norgestrel butanoate, levo-norgestrel cyclopropylcarboxylate levo-norgestrel cyclobutylcarboxylate levo-norgestrel cyclopentylcarboxylate 3-oxime, or levo-norgestrei cyclohexylcarboxylate 3-oxime

The esters may be formulated in a pharmaceutically acceptable aqueous medium, preferably in the form of a microcrystalline suspension of particle size in the range 3 to 10 µ expressed as the 50% cumulative oversize in the Coulter distribution curve. A suitable human dosage 15 contains from 1 to 50 mgms of the ester.

BEST AVAILABLE COPY

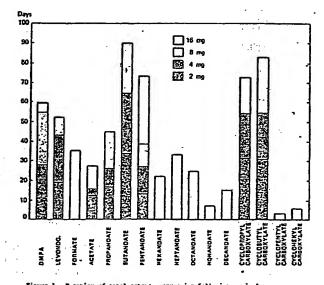


Figure 1. Duration of total extrus suppression following a single . subtutaneous injection of levenorgestrel or its extern to rate.

## CONTRACEPTIVE COMPOSITIONS BASED ON ESTERS OF LEVO-NORGESTREL

This invention relates to injectable therapeutic compositions based on certain esters of levonorgestrel which have valuable long-acting contraceptive properties.

The pharmacological effects of steroids such as levo-norgestrel, progesterone, nortestosterone and norethisterone have been known for very many years.

10 However, their effects as contraceptive agents are relatively short lasting with the consequence that injections at frequent intervals would be required to maintain fertility control.

Attempts have been made to extend the effective

term of contraceptive activity by esterifying the steroids,

a very desirable term being three months or more. However,

to date, only two injectable contraceptive preparations

have been made available. These are depot-medoxy-progesterone

acetate (DMPA), which initially has to be injected at 2-monthly

intervals, and norethisterone enanthate. Both of these suffer from several disadvantages one of them being that although they were developed in the 1960s they have still not been approved by the US Food and Drug Administration.

There is a great need, particularly in developing countries, for an injectable contraceptive which is safe and is effective for a longer period.

With this in mind, over 200 new derivatives of synthetic steroids known to be efficient and "safe" as contra10 ceptive agents have been selected and synthesised. The derivatives were 17-esters of acids of varying chain lengths, nature and degree of unsaturation, (double and/or triple bonds, dienes, enynes, allenes), ring size (cyclopropane to cyclohexane, with or without substitution). Acids
15 containing benzene and furan rings as well as some unusual naturally occurring acids were employed. Oximes of several of the esters and few ethers containing alkyl, aryl and silyl substituents were also prepared.

These were subjected to a biological screening

20 programme designed to uncover new and effective sustainedrelease injectable contraceptives, the compounds being
tested as injectable oily solutions or aqueous suspensions.

The initial test was to determine the duration of suppression
of estrus in female rats, with DMPA and norethisterone

25 enanthate being used as comparison standards. The most

promising compounds were then tested in primates.

In general, the tests indicated a lack of correlation between structure and activity, but it was found that of all the compounds screened in the programme five showed outstanding properties when injected as a micro crystalline suspension in an aqueous medium. The five compounds were levo-norgestrel butanoate, cyclo-propylcarboxylate, cyclobutylcarboxylate, cyclopentylcarboxylate 3-oxime and cyclohexylcarboxylate

It was noted that the duration of action of these five longer acting compounds was also dependent on the nature of the pharmaceutical composition into which they were formulated, even better results being achieved when the compounds were injected as an aqueous suspension of crystals of a particular particle size range. Further, the longer term effectiveness was achieved even with doses smaller than the normal doses of DMPA.

The present invention therefore provides an ester of levo-norgestrel and an aliphatic acid characterized in that the ester is:

levo-norgestrel butanoate

levo-norgestrel cyclopropylcarboxylate

levo-norgestrel cyclobutylcarboxylate

levo-norgestrel cyclopentylcarboxylate 3-oxime

levo-norgestrel cyclohexylcarboxylate 3-oxime

The present invention also provides injectable contraceptive compositions which are suspensions in a pharmaceutically acceptable aqueous medium of one of these selected esters, preferably of microcrystals of particle size in the range 3 to 10,4 expressed as the 50% cumulative oversize in the Coulter distribution curve.

Preferably, the composition is in a unit dosage form containing 1 to 50 mgms of the levo-norgestrel ester.

In general, it is desirable that any administration of steroids is in the smallest dose effective for the purpose and it is a feature of this invention that the compositions if injected at intervals appropriate for DMPA can contain a smaller amount of the steroid, and further, that the dose may still be smaller

5

15

even when the injections are at longer intervals of say, 3 months.

For example, the long-acting contraceptive properties of levo-norgestrel 17 -cyclobutylcarboxylate in aqueous microcrystalline suspension extended beyond a period of 91 days, at a dose of 16 mg in the rat, while a microcrystalline suspension of DMPA has an average suppression duration of 63.7 days (59.6 - 71.2) and an oily solution of norethisterone enanthate an average 10 duration of 23.1 days (13.5 - 34.8). More importantly, the cyclobutylcarboxylate at a dose of 8 mg in the rat gives the same inhibition as DMPA at 16 mg. provided on the pharmacokinetics of levo-norgestrel cyclobutylcarboxylate in the monkey Macaca mulata showed 15 that measurable levels of levo norgestrel are still present in the circulation longer than 100 days after injection.

The invention also provides therefore, a method of contraception wherein there is injected into 20 a human female a dose of an above-mentioned composition

norgestrel esters of particle size in the range 3 to 10, the dose containing from 1 to 50 mgms of the ester.

Preferably, the method of contraception is a 5 continuous one, the injections being given at intervals of from two to six months, though injections can be given at one month intervals if necessary. A suitable dose at one month intervals is from 1 to 6mgms, and at six month intervals is from 20 to 50 mgms.

By levo-norgestrel is meant D-(-)-13 $\beta$  -ethyl-17 $\lambda$ -ethynyl-17 $\beta$  -hydroxygon-4-en-3-one.

The esters used in this invention are prepared by conventional methods as illustrated by the following:

In a preparation of levo-norgestrel cyclobutylcarboxylate thallous ethoxide (0.036 mol) was added to
a solution of levo-norgestrel (0.03 mol) in 100 ml of dry
benzene. The benzene was distilled off slowly but the
volume of the reaction mixture was maintained at about 100
20ml by the dropwise addition of dry benzene thereto. After
200 ml of the benzene were distilled off, the reaction
mixture was cooled in an ice water bath. Thereafter 0.039
mol of cyclobutane carboxylic acid, in its acyl chloride

- 7 -

form, was added dropwise to the reaction mixture which was then refluxed for 5 hours. After cooling, the reaction mixture filtered through a bed of kieselguhr and the resulting precipitate was washed five times with 20 ml of benzene. The solvent was then evaporated and the resulting residue was purified in accordance with conventional procedures.

Although this process gives a useful yield it is difficult to avoid traces of thallium in the final esters.

- In an alternative process, trifluoroacetic 10 anhydride (13 ml = 19.33g, 92.03 ml = mM) and cyclobutane carboxylic acid (8.0 ml = 8.37g, 83.60 mM, freshly distilled) were dissolved in benzene (240 ml) and stirred under anhydrous conditions for 30 minutes. Levo-norgestrel 15 (15g, 48.00mM) was then added and the solution was stirred The reaction mixture at room temperature for 40 minutes. was then diluted with ice water and extracted with ether. The ether extract was washed with saturated sodium bicarbonate solution (2X), water (2X) and brine (1X), dried (Na<sub>2</sub>SO<sub>4</sub>) 20 filtered, and concentrated in vacuo. The residue was then dissolved in acetone and an equal volume of hexane was added to it. Crystallization took place when it was set aside, yielding 10.5g of the above-identified ester having a
  - In a further alternative process the steroid

melting point of 229 - 232°C.

may be esterified by reaction with the acid chloride in the presence of excess acid and a small amount of pyridine, or by using 4-dimethylamino pyridine as a coupling agent.

The corresponding cyclopropyl carboxylate

(M.Pt. 211-4°C) and butyrate ester (M.Pt. 214-9°C) are made by analogous methods.

The oximes are prepared by reacting the appropriate ester with hydroxylamine hydrochloride.

For example, levo-norgestrel cyclopentylcarboxylate

3-oxime was prepared as follows:

Levo-norgestrel 17\$\textsup -cyclopentyl carboxylate (588 mg) was dissolved in pyridine (6 ml) and hydroxylamine-hydrochloride (550 mg) added and the mixture heated at 100°C for 7 minutes and cooled in an ice-bath. Hydrochloric acid (1\textsup , 45 ml) was added dropwise and the mixture extracted with ethyl acetate. The ethyl acetate extract was washed with 1\textsup HCl, water, saturated sodium bicarbonate solution, water and brine and dried over sodium sulphate. Evaporation of the solvent and crystallization of the coxime (593 mg), mp 172.5-174°C; HPLC (\(\mu\)Porasil, heptane: isopropanol 96:4, 2 ml/min, U.V. 254 nm, anti isomer: syn isomer 53:47).

A similar preparation of levo-norgestrel 17 -cyclo-25 hexyl carboxylate 3-oxime gave the oxime (2.2 g) mp

10

167-169°C as a mixture of the <u>anti</u> and <u>syn</u> isomers (53:47 by HPLC - Porasil column, heptane: isopropanol 96:4.

The microcrystalline suspensions of the steroid esters were made by milling the appropriate amount of the ester in an aqueous medium consisting of:

	Benzyl alcohol	1.000% w/v
	Sodium carboxymethyl cellulose	0.500% w/v
	Disodium hydrogen phosphate dihydrate	0.376% w/v
10	Sodium dihydrogen phosphate dihydrate	1.495% w/v.
	Polysorbate 80	0.200% w/v
	Water for injection to	100.000%

The milling was carried out in a Glen Creston

15 M270 micronising ball mill equipped with an agate container and ball. A milling period of up to an hour was required to reduce the particles of ester to the required size.

The bioassay used to determine the duration of action of the contraceptive compositions involved measuring the suppression of estrus in female rats. The compositions were used at four dosage levels, i.e. 2, 4, 8 and 16 mg/rat.

Figure 1 of the accompanying drawings, shows
the effect of various dosages of the compositions based on
three of the selected esters together with the effect of

similar compositions based on esters closely related to those used in the three compositions of this invention.

It will be noted that the three compositions based on the butanoate, cyclopropyl carboxylate and cyclobutyl carboxylate esters suppress estrus to a highly unexpected degree and even at a dosage of 8 mg/rat give better suppression than DMPA at a dosage of 16 mg/rat.

Figure II of the accompanying drawings
illustrates the effect of the cyclobutyl carboxylate ester
lowhen formulated in different types of compositions. In
Figure II the percentage of rats which returned to estrus
(tested in groups of ten) is plotted against weeks and the
symbols used are as follows:

- 0 aqueous microcrystalline suspension
  15 of ester of particle size 3 to 10 as
  hereinbefore defined

  - abla solution in ethyl oleate

The plots illustrate the longer term effect

20 of the ester when micro-dispersed in aqueous medium.

Similar results are obtained using the other two esters

used in this invention.

A more wide-ranging comparison of the butanoate, cyclopropylcarboxylate and cyclobutylcarboxylate esters with other esters of levo-norgestrel is given in the following Table I. Depending on the physical nature of the ester, formulation was either as a microcrystalline suspension in an aqueous vehicle or as a solution in ethyl oleate. The composition of the aqueous vehicle was as given above.

The long-acting properties of each formulation were determined in estrus suppression assay employing virgin, mature (180-200 g) cycling rats of the Sprague-Dawley strain purchased from Charles River Laboratories, USA. Upon receipt the animals were smeared daily for cyclicity and only those animals showing two consecutive normal cycles were used.

- 15 Each animal was injected subcutaneously with 0.8 ml of the test preparation on the same day regardless of the stage of the cycle. Each compound was initially tested on ten rats.

  A series of initial tests with injections of 16 mg/animal were followed by studies at dose levels of 8, 4 and 2 mg/animal
- 20 where necessary. An aqueous microcrystalline suspension of medroxyprogesterone acetate was used as standard. Daily smears were taken starting on the day after injection and continued until such time that cornification of vaginal epithelium was observed and cycling was re-established.
- 25 Duration of cornification suppression is expressed as the

total number of days minus two. The appearance of cornification was not always followed by the return of normal estrus cycles. The majority of experiments were terminated at 91 days post injection even though some of the animals had not exhibited cornified smears.

Numerous estrus suppression assays with an aqueousmicrocrystalline suspension of medroxyprogesterone acetate
at different doses have produced the following durations of
suppression: at 2 mg/rat - 20.9 days, at 4 mg/rat - 28.2

10 days, at 8 mg/rat - 55.5 days, and at 16 mg/rat - 60.0 days.

At the most commonly employed dose, 8 mg/rat, the duration
of activity of the standard varied from 35.1 + 2.0 to 73.2 +

3.4. At 16 mg/rat the suppression of estrus was much
more uniform ranging from 59.6 + 2.4 days to 71.2 + 9.0.

15 Data obtained in seven different experiments

utilizing the cyclobutylcarboxylic ester of levo-norgestrel
indicate a more linear duration of suppression of estrus.

At 2 mg/rat, estrus suppression was obtained for 31.9 days,
at 4 mg/rat it was 55.7 days and at 8 mg/rat it was 89.2 days.

20 The duration of suppression at 8 mg/rat was unrealistically
low due to the arbitrary termination of majority of the
experiments at 91 days post injection. In two separate
experiments which were continued until all animals exhibited
a cornified smear, the average durations of suppression were

25 103.3 + 9.3 and 124.3 + 9.2 days. With the cyclopropyl-

carboxylate the equivalent data is as follows: at 4 mg/rat - 55.6 days and at 8 mg/rat, - 74.0 days. With the butanoate it was at 2 mg/rat, - 34.4 days, at 4 mg/rat, - 65.1 days and at 8 mg/rat, <91 days.

In the series shown in Table 1 and Fig.1, several interesting structure activity relationships can be observed.

Among the straight chain fatty acid esters prepared as microcrystalline suspensions, activity increased sharply from formate to butanoate and then declined just as rapidly.

10 Introduction of a methyl group at C2 of propanoate, at C3 of butanoate and at C4 of pentanoate resulted in marked decreases in activity. Among the cyclic esters, cyclopropylcarboxylate and cyclobutylcarboxylate were longacting whereas cyclopentylcarboxylate and cyclohexylcarboxylate 15 were short acting.

In the series, the influence of formulation on the duration of action was very pronounced with some of the esters and lacking with others. This is illustrated in Table 11. The cyclobutylcarboxylic ester prepared as a 20 microcrystalline suspension in ethyl oleate had only approximately 40% of the activity of the aqueous suspension. Further reduction in activity was observed when this compound was prepared as a solution in a 50:50 mixture of ethyl oleate and benzyl benzoate (BB). Similar differences in activity

were observed among formulations having different particle size distributions; microcrystalline suspensions being more active than the coarser preparations.

A similar series of tests using formulations based on levo-norgestrel ester oximes also produced unexpected results as illustrated in Fig. 111.

Most were tested as aqueous microcrystalline suspensions of the oxime though in the cases of the oximes of levo-norgestrel butanoate, levo-norgestrel isobutanoate, levo-norgestrel cyclopropylcarboxylate and levo-norgestrel itself, the ester oxime could not be formulated as an aqueous microcrystalline suspension and had to be formulated as a solution in ethyl oleate.

Although the cyclopentylcarboxylate and cyclohexyl
carboxylate esters are relatively ineffective even at 16 mg

dose, it was found that their oximes at an 8 mg dose were

comparable with DMPA, which in this series gave lower results

than in the first series. The two oximes also were

surprisingly more effective than closely related ester

oximes such as levo-norgestrel cyclobutylcarboxylate 3-oxime

and levo-norgestrel-(4\*-methyl)cyclohexylcarboxylate 3-oxime.

Again, it was noted that the cyclopentyl and cyclohexyl carboxylate 3-oximes gave better results when formulated as an aqueous microcrystalline suspension of particle size 3 to 10 expressed as the 50% cumulative oversize in the Coulter distribution curve.

It has been observed that progestogens themselves

give rise to some endometrial bleeding between injections,

but the tests on the preferred oximes indicate that they

have a reduced tendency to cause such bleeding and so are

more attractive.

In the accompanying drawings, Fig.III formulation 1

10 was based on depot-medoxy-progesterone-acetate and the
following formulations were based on oximes of

- 2. levo-norgestrel
- levo-norgestrel butanoate
- 4. levo-norgestrel isobutanoate
- 15 5. levo-norgestrel isovalerate
  - 6. levo-norgestrel cyclopropylcarboxylate
  - 7. levo-norgestrel cyclobutylcarboxylate
  - 8. levo-norgestrel (3'-methyl) cyclohexylcarboxylate
  - levo-norgestrel (3'-ethyl) cyclohexylcarboxylate
- 20 10. levo-norgestrel cyclopentylcarboxylate
  - 11. levo-norgestrel cyclohexylcarboxylate
  - 12. levo-norgestrel (4'-methyl) cyclohexylcarboxylate
  - 13. levo-norgestrel acetate

Table I Effect of levonorgestrel esters on the duration of estrus suppression in mature cycling rats.

	· · · · · · · · · · · · · · · · · · ·			<del></del>
ESTER		Dose		Duration of
Compound	Structure	per	Vehicle	suppression in
1		rat	•	days
		mg		mean + SD
	1			
levonorgestrel	н	2	ΑQ	19.5 <u>+</u> 1.6
	•	4	ΑQ	43.7 + 2.7
		8	AQ	52.9 ± 3.9
	C-H			20.2.2.2
formate	с-н	16	ρA	29.7 <u>+</u> 2.2
acetate	٥ !	2	QA	13.1 + 0.9
	٧	4	QA.	16.1 + 1.8
	·· · ·	8	QA.	27.9 <del>+</del> 1.5
1			-	
propanoate	0	2	ΑQ	17.3 + 1.1
		4	AQ	26.5 + 1.6
	, ,	8	AQ	45.6 + 2.5
	0	ļ		
2-methylpropano-		16	AQ	5.7 ± 2.7
ate		1		
	lo			
butanoate	lu a	2	AQ	34.4 + 2.8
		4	AQ	$65.1 \pm 5.9$
•	ю .	-8	ΦQ	91
(0) 0				25.39
(S)-2-methy1-	l i	16	AQ.	3.5 ± 1.2
butanoate	0	ľ		
6		.16	QA.	3.1 + 0.7
2-ethylbutanoate		.10	I AV	3.1 + 0.7
3-methylbutanoate	10	16	ρA	40.3 + 2.5
2-mern's tour amount	1 Per	16	QA.	42.7 + 1.8
	' • `	16	QA.	53.4 + 3.1
	10			-
3,3-dimethy1-	100	16	- AQ	3.7 + 1.3
butanoate				
,	1 10			
pentanoate		2	AQ	13.6 <u>+</u> 1.5
		4	AQ	27.7 + 3.0
	[ ]	8	AQ	39.3 + 3.3
		16	AQ	74.3 + 15.6
	10			
5-methylpentano-	١١٧٨٨	16	QA	50.3 ± 4.7
ate	$I_i \vee Y$			

Table I (Cont'd.)

ESTER Compound	Structure	Dose	Vehicle	Duration of suppression in
		rat		• -
•		mg		days
			,	mean + SD
hexanoate	٥	16	AQ	22.6 <u>+</u> 1.0
			20	26.29
2-ethylhexanoate		16 16	EO	2.6 ± 0.8 1.2 + 0.4
		10	ΟA	1.2 - 0.4
(S)-4-methyl-	.0	16	, AQ	46.6 + 3.3
hexanoate				_
	=	16	EO	22.7 + 3.7
heptanoate	1 o	16	EO	26.2 + 3.7
		16	EO	31.2 + 2.9
		16	AQ	$34.0 \pm 2.8$
•	jo			
octanoate		16	QA.	25.6 <u>+</u> 1.8
nonanoate		16	EO	7.5 + 2.8
	10			10.0/
3-methylnonanoate		16	EO	1.9 ± 0.4
decanoate		16	EO	15.5 ± 3.5
	0	16		2.5 + 0.6
E-2-methylpent-2-		10	AQ	2.5 - 0.0
enoate	¦o '			
E-penta-2,4-dien-		16	AQ	34.8 ± 0.4
oate	io			
	مما	.   16	QA.	9.2 + 1.5
E,E-2-methylhexa- 2,4-dienoate		10	.nq	7.0
Z,4-dienoate	l io			
E-5-phenylpent-2-	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		EO	$5.7 \pm 0.6$
en-4-ynoate	1 10	16	AQ	$1.6 \pm 0.4$
	II OME	16	ΑQ	29.5 + 1.8
methoxyacetate	10	10	AV	
cyanoacetate	II CN	16	. QA	85.0 <u>+</u> 3.0
cyclopropyl-	10-	2	AQ	44.9 + 2.2
carboxylate	1 1 1	2		29.0 + 1.7
	1 7	4		68.3 + 3.5
		4		$42.8 \pm 2.3$
		8	1 '	74.0 74.1 + 2.3
1		8	QA P	74.1 7 2.3

Table I (Cont'd.)

ESTER		Dose		Duration of
Compound	Structure	per	Vehicle	suppression in
		rat		days .
		. mg		mean + SD
cyclobutylcarboxylate	10	9	40	*
(*average of 6	₩	2 4	AQ AQ	31.9 55.7
assays)	,	6	AQ	93.8
2-methylcyclobutyl-	! O . IL ∕	16		
carboxylate		10	AQ	55.2 <u>+</u> 3.1
	0			•
2-ethylcyclobutyl- carboxylate	,	16	. QA	33.6 <u>+</u> 6.6
Carboxylate	i			
2-hexylcyclobutyl-	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	16	ΑQ	46.1 + 4.4
carboxylate	10		·	
3-cyclobutyl-	L	16	40	20 5 . 0 5
propanoate		10	AQ	39.5 <u>+</u> 2.1
	الأم م			·
5-cyclobutyl- pentanoate	1~~ ~	16	EO	14.4 <u>+</u> 6.8
0				
7-cyclobutyl-	$\sim\sim$	16	EO	3.3 + 1.1
heptanoate	10			<del>-</del> .
cyclopentyl-		16	AQ	3.9 + 1.2
carboxylate				. 1.2
cyclopent-1-				
enylacetate		16	AQ	16.3 <u>+</u> 2.4
	1:0			
cyclohexylcarboxylate		16	QA	6.5 <u>+</u> 2.6
trans-4-methylcyclo-	<u>п</u>	16	AQ	303399
hexylcarboxylate		10	AŲ	19.1 <u>+</u> 2.2
	_ 1			
trans-4-hexylcyclo- lu hexylcarboxylate	$\bigcirc$ $\sim$	16	ρA	1.9 ± 0.3
	ا م ا			
cyclohexylacetate		16	AQ	1.7 <u>+</u> 0.3
cyclohex-1-enyl-	0	16		_
acetate		10	QA	2.9 <u>+</u> 0.3
			,	

Table I (Cont'd.)

ESTER	Dose	Vehicle	Duration of suppression in	
Compound	Structure	rat		days
L	•	mg		mean <u>+</u> SD
5'-methyl-2'-furyl-	ů []	16	EO	41.6 + 5.2
acetate	: \^\\	16	AQ	47.6 ± 3.4
3-(5'-methyl-2'- furyl)propanoate		16	AQ	47.0 <u>+</u> 3.4
3-(5'-ethyl-2'-	:0	16	EO	29.2 + 3.2
furyl)propanoate		16	AQ	21.1 ± 3.5
4-(5'-methyl-2'- furyl)butanoate		16	ΑQ	2.8 <u>+</u> 0.6
adamantylcarbox- ylate		<b>16</b>	. AQ	2.4 <u>+</u> 0.8

Table II Effect of Formulation and Vehicle on Duration of Action of Levonorgestrel Esters

Ester	Formulation	Vehicle	Dose mg/ Rat	Response Days Mean <u>+</u> SD
cyclobutyl- carboxylate (121)	solution suspension suspension suspension suspension suspension	EO + BB EO EO AQ AQ AQ	2 4 8 2 4 8	$   \begin{array}{r}     14.7 + 2.7 \\     24.3 + 1.9 \\     49.9 + 6.7 \\     26.6 + 1.6 \\     66.9 + 2.9 \\     124.3 + 9.2   \end{array} $
butanoate	solution suspension	EO + BB AQ	2 2	19.1 <u>+</u> 2.2 34.4 <u>+</u> 2.8
pentanoate (104)	solution suspension	EO + BB	2 2	1.3 ± 0.3 13.6 ± 1.5
heptanoate (109)	solution suspension	EO. AQ	16 16	$\begin{array}{c} 31.2 \pm 2.9 \\ 34.0 \pm 2.8 \end{array}$
5'-methyl- 2'-furyl) acetate ( <u>135</u> )	solution suspension	EO AQ	16 16	41.6 ± 5.2 47.6 ± 3.4
5'-ethyl- 2'-furyl propanoate (136)	solution suspension	EO AQ	16 16	29.2 <u>+</u> 3.2 21.9 <u>+</u> 3.5
		<u> </u>		

#### CLAIMS:

1. An ester of levo-norgestrel and an aliphatic acid characterized in that the ester is:

levo-norgestrel butanoate,

- levo-norgestrel cyclopropylcarboxylate,

  levo-norgestrel cyclobutylcarboxylate,

  levo-norgestrel cyclopentylcarboxylate 3-oxime, or

  levo-norgestrel cyclohexylcarboxylate 3-oxime.
- An ester of levo-norgestrel and an aliphatic
   acid characterized in that the ester is levo-norgestrel butanoate.
  - 3. An ester of levo-norgestrel and an aliphatic acid characterized in that the ester is levo-norgestrel cyclopropylcarboxylate.
- 15 4. An ester of levo-norgestrel and an aliphatic acid characterized in that the ester is levo-norgestrel cyclopentylcarboxylate 3-oxime.
- An ester of levo-norgestrel and an aliphatic acid characterized in that the ester is levo-norgestrel
   cyclohexylcarboxylate 3-oxime.

- An injectable contraceptive composition which is a suspension of a levo-norgestrel ester in a pharmaceutically acceptable aqueous medium characterized in that the ester is one claimed in any preceding claim.
- A composition as claimed in Claim 6 characterized in that the ester is suspended in the form of microcrystals of particle size in the range 3 to 10 expressed as the 50% cumulative oversize in the Coulter distribution curve.
- 8. A method of contraception wherein there is injected
  10 into a human female a dose of a composition which is an
  aqueous suspension of a levo-norgestrel ester characterized
  in that the composition is as claimed in Claim 6 or 7 and the
  dose contains from 1 to 50 mgms of the ester.

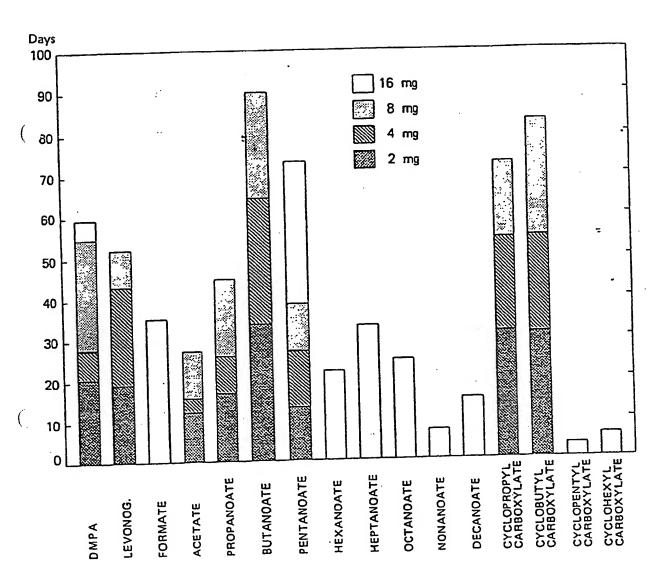
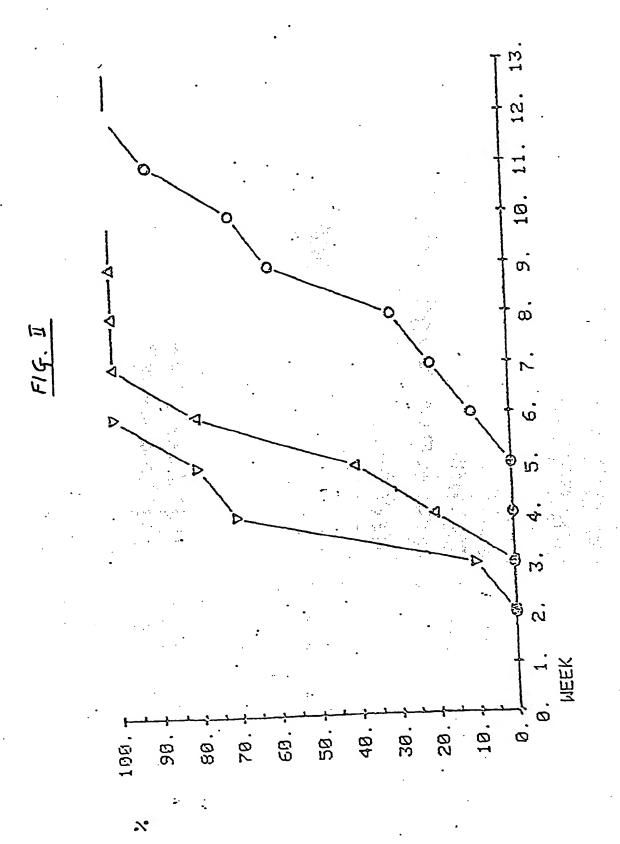


Figure 1. Duration of total estrus suppression following a single subcutaneous injection of levonorgestrel or its esters to rats.

XXXID: <EP\_\_\_0129947A2\_I\_>



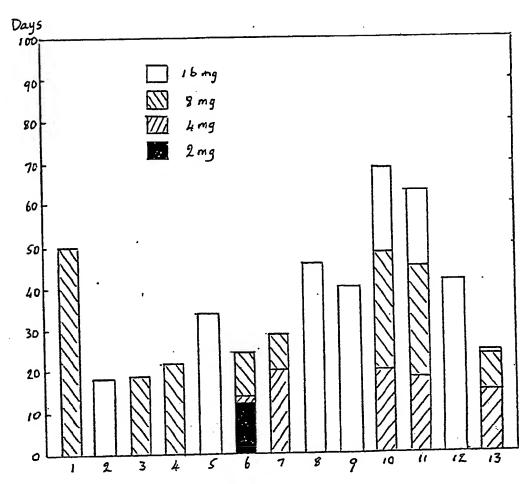


Figure III. Duration of total estrus suppression following a single subcutaneous injection of oximes to rats.

OCID: <EP\_\_0129947A2\_l\_>

1) Publication number:

**0 129 947** A3

12

#### **EUROPEAN PATENT APPLICATION**

- (21) Application number: 84300611.5
- Date of filing: 31.01.84

(a) Int. Cl.4: **C 07 J 1/00**, C 07 J 41/00, A 61 K 31/565

(37) Priority: 19.05.83 GB 8313921

- Date of publication of application: 02.01.85
  Bulletin 85/1
- Designated Contracting States: AT BE CH DE FR GB IT LI LU NL SE
- 88 Date of deferred publication of search report: 22.05.85 Bulletin 85/21

- Applicant: WORLD HEALTH ORGANISATION, CH-1211 Geneva 27 (CH)
- (7) Inventor: Archer, Sidney Department of Chemistry, Renssalaer Polytecnic Institute, Troy New York 12181 (US) Inventor: Diczfalusy, Egon Swedish Medical, Research Council Karolinska Sjukhuset, S-104 01 Stockholm 60 (SE) Inventor: Fried, Josef Department of Chemistry, The University of Chicago 5735 South Ellis Avenue, Chicago

Illinois 60637 (US)
Inventor: Benagiano, Guiseppe, Universita di Roma
Policiinico Umberto 1, I-00161 Rome (IT)
Inventor: Crabbe, Pierre, 11 Rue de l'Abbe Gregoire,
F-75006 Paris (FR)

inventor: Djerassi, Carl, Department of Chemistry Stanford University, Stanford California 94305 (US)

(A) Representative: Arthur, Bryan Edward et al, Withers & Rogers 4 Dyer's Buildings Holborn, London EC1N 2JT (GB)

Contraceptive compositions based on esters of levo-norgestrel.

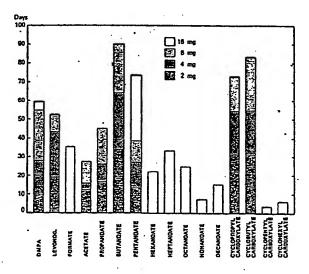
A small group of esters of levo-norgestrel have been found to be unexpectedly effective as long term contraceptive agents. The esters are:

levo-norgestrel butanoate,

levo-norgestrel cyclopropylcarboxylate, levo-norgestrel cyclobutylcarboxylate,

levo-norgestrel cyclopentylcarboxylate 3-oxime, or levo-norgestrel cyclohexylcarboxylate 3-oxime.

The esters may be formulated in a pharmaceutically acceptable aqueous medium, preferably in the form of a microcrystalline suspension of particle size in the range 3 to 10 µ expressed as the 50% cumulative oversize in the Coulter distribution curve. A suitable human dosage contains from 1 to 50 mgms of the ester.



EP 0 129 947 A



#### **EUROPEAN SEARCH REPORT**

0 1 2 9 9 4 7 EP 84 30 0611

	DOCUMENTS CONS				
Category		ith indication, where appropriate evant passages	e,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Ci. 3)
х	STEROIDS, vol. 1983, pages 349 San Francisco ( A. SHAFIEE et a contraceptive a and alicyclic c of levonorgestr  * Pages 351-355	-359, Holden-Daus), l.: "Long-actingents: aliphatarboxylic esterel".	ay, ng ic	1-3, 6-8	C 07 J 1/00 C 07 J 41/00 A 61 K 31/565
			1		
х	Idem, pages 419 G. BIALY et al. contraceptive a activity relatiries of norethiorgestrel ester	: "Long-acting gents: structur onships in a se sterone and lev	re e-	1-3, 6-8	
l	* Pages 419-427	and 435-439 *			TECHNICAL FIELDS
Х	BULLETIN DES SOCIETES CHIMIQUES BELGES, vol. 92, no. 3, March 1983, Brussels (BE), pages 275-287, PIERRE CRABBE et al.: "Long-acting contraceptive agents: X-ray study of levonorgestrel esters".			1-3, 6-8	A 61 K C 07 J
		·			
x	CHEMICAL ABSTRACTS, vol.97, no.21, 22 November 1982, page 860, abstract 182724x (COLUMBUS OHIO, US); R. VLAKHOV et al.: "Synthesis of some esters of norethisterone and d-levonorgestrel as possible fertility regulators"			1,3,	
			-		in the second se
		<del></del> /	/	*	
	The present search report has t	peen drawn up for all claims			**************************************
	Place of search	Date of completion of th	e search	1	Examiner
7	THE HAGUE 05-12-1984			HEN	RY J.C.
Y: par doo A: tecl	CATEGORY OF CITED DOCL ticularly relevant if taken alone ticularly relevant if combined w sument of the same category hnological background inwritten disclosure tremediate document	E: e a arith another D: d L: d	arlier patent fter the filing ocument cit ocument cit	document, I date ed in the app ed for other	ying the invention out published on, or olication reasons

EPO Form 1503. 03.82

15



#### **EUROPEAN SEARCH REPORT**

012947 EP 84 30 0611

Page 2

tegory		th indication, where appropriate, vant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 3)
	& Int. Conf. Che		-	
	* Abstract *			
	FR-A-2 318 645 * Claims 1, 15		4-8	
	-			
				TECHNICAL FIELDS
				SEARCHED (Int. Cl. 3)
:	•			
	The present search report has t	peen drawn up for all claims		
	Place of search .	Date of completion of the search		Examiner
(: pa	CATEGORY OF CITED DOCU	E : earlier pa	itent document, filing date	rlying the invention , but published on, or
r: pa: do	rticularly relevant if combined w cument of the same category chnological background n-written disclosure	rith another D: documer L: documer	nt cited in the ap nt cited for other	pplication reasons

EPO Form 1503. 03 82

# This Page is Inserted by IFW Indexing and Scanning Operations and is not part of the Official Record

#### **BEST AVAILABLE IMAGES**

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images include but are not limited to the items checked:
BLACK BORDERS
☐ IMAGE CUT OFF AT TOP, BOTTOM OR SIDES
☐ FADED TEXT OR DRAWING
☐ BLURRED OR ILLEGIBLE TEXT OR DRAWING
☐ SKEWED/SLANTED IMAGES
☐ COLOR OR BLACK AND WHITE PHOTOGRAPHS
☐ GRAY SCALE DOCUMENTS
LINES OR MARKS ON ORIGINAL DOCUMENT
REFERENCE(S) OR EXHIBIT(S) SUBMITTED ARE POOR QUALITY

### IMAGES ARE BEST AVAILABLE COPY.

OTHER: \_\_\_\_

As rescanning these documents will not correct the image problems checked, please do not report these problems to the IFW Image Problem Mailbox.